

# EarthScope Student Geochronology Research And Training Program Laboratory Overview

University of Cincinnati Terrestrial Cosmogenic Nuclide Laboratories  
March 8, 2015

## Laboratory Description

The University of Cincinnati Terrestrial Cosmogenic Nuclide (TCN) Laboratories were established in 2004 and have processed several thousand samples over the past decade. Research has focused on Quaternary paleoenvironmental change and landscape evolution along active plate margins. This has included dating alluvial fans, strath terraces, river terraces, moraines and landslides, and determining erosion rates using sediments.

The TCN Laboratories have the facilities to prepare sediment and rock samples for  $^{10}\text{Be}$ ,  $^{26}\text{Al}$  and  $^{36}\text{Cl}$  cosmogenic surface exposure dating and erosion studies. This includes: a rock crushing laboratory; a heavy liquid separation laboratory; a chemistry laboratory for  $^{10}\text{Be}$  and  $^{26}\text{Al}$  that has ultrasonic baths, hot rollers for leaching sediment and rock, columns for cation and anion exchange, and HF hoods for acid work; a perchloric laboratory; and a station for target loading. In addition, there is a separate laboratory for  $^{36}\text{Cl}$  work, which has a chemical hood, ovens, centrifuges and balances. The  $^{36}\text{Cl}$  Laboratory is the latest addition to the TCN Laboratories, being established in 2014 it has already processed several dozen samples. A laboratory manager oversees the safety and training activities of the laboratories.

## Expected Time Frame and Procedure

Visitors should discuss the nature of the samples they would like to process before writing grants. Usually, we estimate that it takes about two-three weeks to prepare a batch of about ten samples. But this might be longer dependent upon the type of samples being processed. Generally, for each additional sample over ten then add one extra day should be added to the laboratory visit. Visitors are encouraged to fully discuss the nature of their samples with the laboratory director to help calculate a realistic timeframe for their work.

Visitors will first undertake safety training online and in the laboratory. This is followed by a day-long course on the theory, methods and application of terrestrial cosmogenic nuclides. The general procedure for  $^{10}\text{Be}/^{26}\text{Al}$  work includes crushing and sieving samples to a 250-500  $\mu\text{m}$  particle size and treatment with aqua regia. The sample will then be placed in a 24-hour 5% hydrofluoric solution. The remaining sample was rinsed and agitated with a high velocity 10% Lauryl Amine and  $\text{CO}_2$  spray. Samples were then be subjected to further quartz purification using a Frantz Magnetic Separator and lithium sodiumpolytungstate. The chemical preparation of quartz follows the method developed by Kohl and Nishiizumi (1992, *Geochimica et Cosmochimica Acta*). The resulting  $\text{Be}(\text{OH})_2$  and  $\text{Al}_2\text{O}_3$  gels of each sample will be dried and combusted at  $700^\circ\text{C}$  before being loaded into targets for accelerator mass spectrometry (AMS) at PRIME lab at Purdue University to determine  $^{10}\text{Be}/^9\text{Be}$  and  $^{26}\text{Al}/^{27}\text{Al}$  ratios.

Generally, samples for  $^{36}\text{Cl}$  will be processed for whole-rock analysis as outlined by

Stone et al. (1996, *Geochimica et Cosmochimica Acta*). Most samples will be crushed and sieved to collect 250–500 micron particle size fraction. Crushed samples will be leached thoroughly, first in 18mΩ water and then in 10% HNO<sub>3</sub> for more than 12 h at room temperature. Major elements, including U and Th, before and after leaching, will be determined by X-ray fluorescence, and B and Gd by prompt-gamma-emission spectrometry. The samples will be dissolved over 2 days in a 15M HF and 2M HNO<sub>3</sub> mixture at 60–70°C. A chloride spike (non-terrestrial <sup>37</sup>Cl/<sup>35</sup>Cl) will be added to each dissolved sample. Chloride will be recovered from the sample solutions as AgCl. The <sup>36</sup>Cl/<sup>37</sup>Cl and <sup>35</sup>Cl/<sup>37</sup>Cl will be measured using accelerator mass spectrometry (AMS) at the PRIME Laboratory of Purdue University.

### **Analytical Costs**

Students should budget \$500 for the training fee. Each sample will cost \$400 for processing, which includes all consumables and supplies and use of equipment, plus \$235 per sample for each AMS measurements. If visitors only require <sup>10</sup>Be analysis then each sample will cost \$635, but if they require both <sup>10</sup>Be and <sup>26</sup>Al then the cost will be \$870 per sample. If visitors are undertaking <sup>36</sup>Cl work then they should budget an addition \$300 for bulk chemical analysis (each <sup>36</sup>Cl sample will cost \$935). There are no costs for chemical blanks. We would generally expect visitors to processes a minimum of ten samples to make the training and visit worthwhile.

### **Preparation for the Visit**

No sample preparation is required before visiting the TCN Laboratories. But visitors will be required to fully discuss their projects and nature of their samples before they write a grant proposal. Visitors will be encouraged to take the online safety courses before arriving.

### **Date Processing and Interpretation**

Visitors will be trained in how to process and reduce their data to calculate ages and/or erosion rates.

### **Expected Laboratory Availability**

The TCN Laboratories are available most of the year round, but we would appreciate a lead-time of about three months.

### **Laboratory Staff**

Professor Lewis Owen ([lewis.owen@uc.edu](mailto:lewis.owen@uc.edu)) is the director of the TCN Laboratories and has over 15 years of experience managing geochronology laboratories. Ms. Sarah Hammer ([tritscsh@ucmail.uc.edu](mailto:tritscsh@ucmail.uc.edu)) is the manager and safety officer. She joined the TCN Laboratories in 2011. Assistant Professor Dylan Ward co-directs with Lewis Owen the <sup>36</sup>Cl laboratory.

### **Contacts**

Please contact Lewis Owen ([lewis.owen@uc.edu](mailto:lewis.owen@uc.edu)) if you are interested in collaborative work and preparing samples for <sup>10</sup>Be, <sup>26</sup>Al and/or <sup>36</sup>Cl.